metal-organic papers

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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ R factor = 0.034 wR factor = 0.082 Data-to-parameter ratio = 13.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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catena-Poly[[[diaqua(2,2'-diamino-4,4'-bithiazole- $\kappa^2 N, N'$)manganese(II)]- μ -terephthalato- $\kappa^2 O:O'$] dihydrate]

The title Mn^{II} complex, $[Mn(C_8H_4O_4)(C_6H_6N_4S_2)(H_2O)_2]$ -2H₂O, assumes a distorted octahedral coordination geometry formed by 2,2'-diamino-4,4'-bithiazole (DABT), terephthalate dianions and water molecules. The terephthalate dianions, each located on an inversion center, bridge Mn^{II} ions, forming polymeric complex chains. Additionally, π - π stacking is observed between parallel bithiazole ligands.

Comment

Transition metal complexes of diaminobithiazole (DABT) have shown potential application in some fields, for example, a Co^{II} complex and a Ni_{II} complex with the DABT ligand have been found to be effective inhibitors of DNA synthesis of the tumor cells (Waring, 1981; Fisher *et al.*, 1985). As part of ongoing investigations on metal complexes incorporating the DABT ligand (Liu *et al.*, 2001), the title Mn^{II} complex, (I), has been prepared in the laboratory and its X-ray crystallogaphic structure is reported here.



The structure of (I) is shown in Fig. 1. The Mn^{II} complex assumes a distorted octahedral coordination (Table 1), formed by a DABT ligand, two terephthalate dianions and two water molecules. The terephthalate dianions, each located on an inversion center, bridge Mn^{II} ions, forming polymeric complex chains (Fig. 1).

The two thiazole rings of the DABT ligand are coplanar, the maximum atomic deviation from the mean plane being 0.0379 (14) Å (for S1). The dihedral angle between the two thiazole planes is 2.60 (10)°, which agrees with 4.57 (7)° found in [Mn(DABT)(oxydiacetate)] (Luo *et al.*, 2004), but different from 20.02 (8)° in [Ni(DABT)(iminodiacetate)] (Liu & Xu, 2005).



Figure 1

Part of the polymeric structure of (I), shown with 40% probability displacement ellipsoids (arbitrary spheres for H atoms). Dashed lines indicate hydrogen bonds. [Symmetry codes: (vi) 1 - x, 1 - y, 1 - z; (vii) 2 - x, 3 - y, 1 - z.]



Figure 2

Partially overlapped arrangement of DABT ligands, showing $\pi - \pi$ stacking [symmetry code: (viii) 1 - x, 1 - y, -z].

A partially overlapped arrangement of the parallel thiazole rings is observed (Fig. 2). The face-to-face separation between mean planes of the S1-bithiazole and S1^{viii}-bithiazole ligands is 3.319 (7) Å [symmetry code: (viii) 1 - x, 1 - y, -z], suggesting the existence of π - π stacking.

An extensive $N-H \cdots O$ and $O-H \cdots O$ hydrogen-bonding network is observed (Table 2), which stabilizes the crystal structure.

Experimental

An aqueous solution (20 ml) containing DABT (0.20 g, 1 mmol) and MnCl₂·2H₂O (0.16 g, 1 mmol) was mixed with an aqueous solution (10 ml) of terephthalic acid (0.17 g, 2 mmol) and NaOH (0.08 g, 2 mmol). The mixture was refluxed for 6 h. The solution was filtered after cooling to room temperature. Single crystals of (I) were obtained from the filtrate after 3 d.

Crystal data

$[Mn(C_8H_4O_4)(C_6H_6N_4S_2)-$	$\gamma = 111.332 \ (14)^{\circ}$
$(H_2O)_2]\cdot 2H_2O$	V = 1000.9 (3) Å ³
$M_r = 489.38$	Z = 2
Triclinic, P1	$D_x = 1.624 \text{ Mg m}^{-3}$
a = 10.0666 (12) Å	Mo $K\alpha$ radiation
b = 10.2989 (14) Å	$\mu = 0.92 \text{ mm}^{-1}$
c = 11.5496 (15) Å	T = 295 (2) K
$\alpha = 111.146 \ (13)^{\circ}$	Prism, yellow
$\beta = 96.060 \ (11)^{\circ}$	$0.25 \times 0.18 \times 0.15 \text{ mm}$

5305 measured reflections

 $R_{\rm int}=0.015$

 $\theta_{\rm max} = 25.2^{\circ}$

3550 independent reflections 3085 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0346P)^2]$

+ 0.6656P] where $P = (F_0^2 + 2F_c^2)/3$

 $\Delta \rho_{\rm min} = -0.34 \text{ e } \text{\AA}^{-3}$

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.39 \text{ e} \text{ Å}^{-3}$

Data collection

Rigaku R-AXIS RAPID diffractometer ω scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{\min} = 0.766, \ T_{\max} = 0.870$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.034$ wR(F²) = 0.082 S = 1.033550 reflections 262 parameters H-atom parameters constrained

Table 1

Selected bond lengths (Å).

Mn-N1	2.2391 (19)	Mn-O2	2.2010 (18)
Mn-N2	2.252 (2)	Mn-O11	2.1470 (17)
Mn-O1	2.2080 (17)	Mn-O21	2.2049 (16)

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
N3−H3A…O1	0.82	2.36	3.061 (4)	143
$N3 - H3B \cdots O1W^{i}$	0.82	2.14	2.942 (4)	170
$N4-H4A\cdots O21$	0.90	2.34	3.016 (4)	132
$N4-H4B\cdots O12^{ii}$	0.84	2.14	2.959 (4)	166
$O1-H1A\cdots O21^{iii}$	0.84	2.15	2.959 (2)	161
$O1 - H1B \cdots O12^{iii}$	0.82	1.93	2.738 (3)	170
$O2-H2A\cdots O22$	0.91	1.75	2.641 (3)	165
$O2 - H2B \cdots O1W^{iv}$	0.84	2.02	2.839 (3)	167
$O1W-H1C\cdots O22$	0.88	1.89	2.762 (3)	170
$O1W - H1D \cdots O2W^{v}$	0.90	1.83	2.716 (4)	170
$O2W - H2C \cdots O12$	0.94	1.88	2.765 (4)	157

Symmetry codes: (i) x, y - 1, z;(ii) -x + 1, -v + 2, -z + 1;(iii) -x + 2, -y + 2, -z + 1; (iv) -x + 2, -y + 2, -z; (v) x, y, z - 1.

Amino H atoms and water H atoms were located in a difference Fourier map and refined as riding in their as-found relative positions, with $U_{iso}(H) = 1.2U_{eq}(N,O)$. Other H atoms were placed in calculated positions, with C-H = 0.93 Å, and refined as riding, with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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